

## COURSE OUTLINE: MCH256 - INTRO THERMO DYNAMIC

Prepared: Howard Gray Approved: Corey Meunier, Chair, Technology and Skilled Trades

Course Coder Title					
Course Code: Title	MCH256: INTRODUCTORY THERMO DYNAMICS				
Program Number: Name	4039: MECH. ENG. TN-MANUFA				
Department:	MECHANICAL TECHNIQUES PS				
Semesters/Terms:	22W				
Course Description:	The general objective of this course is to give students destined for the mechanical trades an introduction to thermodynamics. The course covers temperature, pressure, volume relationships for gases, specific heat, the relationship between heat and work, heat engines and heat transfer				
Total Credits:	3				
Hours/Week:	3				
Total Hours:	45				
Prerequisites:	There are no pre-requisites for this course.				
Corequisites:	There are no co-requisites for this course.				
Substitutes:	MCH130				
Vocational Learning	4039 - MECH. ENG. TN-MANUFA				
Outcomes (VLO's) addressed in this course:	VLO 1 Complete all work in compliance with current legislation, standards, regulations and guidelines.				
Please refer to program web page for a complete listing of program	VLO 2 Apply quality control and quality assurance procedures to meet organizational standards and requirements.				
outcomes where applicable.	VLO 3 Comply with current health and safety legislation, as well as organizational practices and procedures.				
	VLO 4 Apply sustainability best practices in workplaces.				
	VLO 5 Use current and emerging technologies to support the implementation of mechanical engineering projects.				
	LO 6 Analyze and solve mechanical problems by applying mathematics and fundamentals of mechanical engineering.				
	VLO 10 Verify the specifications of materials, processes and operations to support the design and production of mechanical components.				
	VLO 12 Develop strategies for ongoing personal and professional development to enhance work performance.				
Essential Employability Skills (EES) addressed in	EES 1 Communicate clearly, concisely and correctly in the written, spoken, and visual form that fulfills the purpose and meets the needs of the audience.				
this course:	EES 2 Respond to written, spoken, or visual messages in a manner that ensures effective communication.				

In response to public health requirements pertaining to the COVID19 pandemic, course delivery and assessment traditionally delivered in-class, may occur remotely either in whole or in part in the 2021-2022 academic year.

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	EES 3	Execute mathemati	ical operations accurately.
	EES 4		approach to solve problems.
	EES 5	,	nking skills to anticipate and solve problems.
	EES 6		anize, and document information using appropriate technology
	EES 7	,	and apply relevant information from a variety of sources.
	EES 8	-	e diverse opinions, values, belief systems, and contributions of
	EES 9		in groups or teams that contribute to effective working ne achievement of goals.
	EES 10	Manage the use of	time and other resources to complete projects.
	EES 11	Take responsibility	for ones own actions, decisions, and consequences.
General Education Themes:	Science	and Technology	
Course Evaluation:	Passing	Grade: 50%, D	
	A minimu for gradu		2.0 or higher where program specific standards exist is required
Other Course Evaluation & Assessment Requirements:	A+ 90 - 1 A 80 - 89 B 70 - 79 C 60 - 69 D 50 - 59 F (Fail) 4 CR (Cree S Satisfa U Unsati X A temp additiona NR Grad	% 3.00   % 2.00   % 1.00   9% and below 0.00   dit) Credit for diploma   ctory achievement in   sfactory achievement in   sfactory grade limited t   limite to complete th   e not reported to Reg	a requirements has been awarded. field /clinical placement or non-graded subject area. t in field/clinical placement or non-graded subject area. o situations with extenuating circumstances giving a student e requirements for a course.
Books and Required Resources:	Hand out Hand out	ts is provided by instruc	stor
Course Outcomes and	Course	Outcome 1	Learning Objectives for Course Outcome 1
Learning Objectives:		vior of Gases	Potential Elements of the Performance: 1.1 Demonstrate an understanding of perfect gases under the following conditions: a. Constant temperature (Boyle`s Law) b. Constant volume (Charles Law) c. Constant pressure (Gay-Lussac`s Law) d. Varying temperature, volume and pressure (General Gas Law)

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Course Outcome 2	Learning Objectives for Course Outcome 2
2. Specific Heat	Potential Elements of the Performance: 2.1 Define and calculate specific heats under conditions of constant volume and constant pressure.
Course Outcome 3	Learning Objectives for Course Outcome 3
3. 1st and 2nd Laws of Thermodynamics	Potential Elements of the Performance: 3.1 Explain the concept of heat 3.2 Explain the first and second law of thermodynamics to demonstrate an understanding of the relationship between heat, energy and work.
Course Outcome 4	Learning Objectives for Course Outcome 4
4. Work	Potential Elements of the Performance: 4.1 Calculate the work done under the following conditions: a. Constant pressure b. Constant Temperature c. Adiabatic expansion and compression d. Polytrophic Compression and expansion
Course Outcome 5	Learning Objectives for Course Outcome 5
5. Practical Thermodynamic Cycle - Heat Engines	Potential Elements of the Performance: 5.1 Explain the concept of a heat engine. 5.2 Identify the application for each of the following thermodynamic cycles: a. Carnot b. Rankine c. Otto d. Diesel e. Brayton
Course Outcome 6	Learning Objectives for Course Outcome 6
6. Heat Transfer	Potential Elements of the Performance: 6.1 Explain the concepts of heat transfer through conduction convection and radiation. 6.2 Utilize the equations for heat transfer.

Evaluation Process and Grading System:	Evaluation Type	Evaluation Weight		
Grading System.	Tests	100%		
Date:	September 3, 2021			
Addendum:	Please refer to the information.	course outline adder		

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